MATERIAL SAFETY DATA SHEET

SRM Supplier: National Institute of Standards and Technology

Standard Reference Materials Program 100 Bureau Drive, Mail Stop 2320 Gaithersburg, Maryland 20899-2320 RM Number: 8640 MSDS Number: 8640

RM Name: Microspheres with Immobilized

Fluorescein Isothiocyanate

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SECTION I. MATERIAL IDENTIFICATION

Material Name: Microspheres with Immobilized Fluorescein Isothiocyanate

Description: This material consists of six sealed plastic bottles: five containing a suspension of microspheres with different amounts of labeled fluorescein isothiocyanate; one bottle containing a suspension of microspheres without immobilized fluorescein isothiocyanate. The suspension has a nominal microsphere concentration of 10⁶ particles/mL. Each individual plastic bottle contains approximately 2.0 mL of the suspension.

The suspension buffer is a PBS buffer containing 15 mM sodium azide and 0.1 % tween 20.

Other Designations: Microspheres with Immobilized Fluorescein Isothiocyanate (FITC; fluorescein, 5-isothiocyanato; fluorescein 5-isothiocyanate; 5-isothiocyanatofluorescein; 5-fluorescein isothiocyanate) in PBS buffer (phosphate buffered saline) containing 15 mM Sodium Azide (hydrozoic acid, sodium salt) and 0.1% Tween 20 (polyoxyethylene [20] sorbitan monolaurate; polysorbate 20; ethoxylated sorbitan monolaurate).

Information on Ingredients:

Name	Chemical Formula	CAS Registry Number	
Fluorescein Isothicyanate Labeled Microspheres:			
Flourescein Isothiocyanate	$C_{21}H_{11}NO_5S$	3326-32-7	
PBS buffer:			
Disodium Hydrogen Phosphate	Na_2HPO_4	7558-79-4	
Sodium Chloride	NaCl	7647-14-5	
Sodium Azide	NaN_3	26628-22-8	
Tween 20	$C_{58}H_{114}O_{26}$	9005-64-5	

DOT Classification: Not regulated by DOT

MSDS 8640 Page 1 of 4

SECTION II. HAZARDOUS INGREDIENTS

Hazardous Component	Nominal Concentration (%)	Exposure Limits and Toxicity Data
Sodium Azide ^a	0.01	OSHA ceiling (skin): 0.3 mg/m ³
		ACGIH ceiling: 0.3 mg/m ³
		Human, Oral TD _{Lo} : 710 μg/kg
Tween 20 ^a	0.1	No occupational exposure limits established.
Microspheres with Immobilized Fluorescein Isothiocyanate ^b		No occupational exposure limits established
		No toxicity data exists. The toxic properties have not been thoroughly investigated.

^aExposure limits and toxicity data provided are for the concentrated forms of sodium azide and tween 20. Their concentrations in this buffer, however, are very low, and it may be slightly irritating.

SECTION III. PHYSICAL/CHEMICAL CHARACTERISTICS

Microspheres with Immobilized Fluorescein Isothiocyanate			
Appearance and Odor: colorless, odorless, clear liquid suspension			
Density: ~1.05 g/cm ³			
Boiling Point: 100 °C			
Solubility in Water: highly soluble (water-based)			

SECTION IV. FIRE AND EXPLOSION HAZARD DATA

Flash Point: Not Applicable Method Used: Not Applicable Autoignition Temperature: Not Applicable

Flammability Limits in Air (Volume %): UPPER: Not Applicable LOWER: Not Applicable

Unusual Fire and Explosion Hazards: The suspended material is not flammable. Sodium azide is a slight fire hazard and a severe explosion hazard. Sodium azide reacts with many heavy metals to form explosive compounds. Sodium azide also reacts with metal halides to give metal azide halides, many of which are explosive.

Extinguishing Media: Use extinguishing agents appropriate for surrounding fire.

Special Fire Procedures: Fire fighters should wear a self-contained breathing apparatus (SCBA) with a full-face piece in the pressure demand or positive mode and other protective clothing.

MSDS 8640 Page 2 of 4

^bThe chemical, physical and toxic properties of this product, microspheres with immobilized fluorescein isothiocyante, have not been thoroughly investigated; however, the organic fluorescent dye, flourescein isothiocyanate, is a suspected carcinogen.

SECT	TION V. REACTIVIT	гу D ата						
S	Sodium Azide:							
	Stability:	Stab	ole	X	_ Unstable			
	Sodium azide may (Materials to Avoid		eated. Sodium	azide may	also form exp	losive compo	unds. See "Incor	npabilities
	Conditions to Avoid: Avoid heat, flames, sparks, and other sources of ignition. Avoid friction or contamination. See "Incompatibilities (Materials to Avoid)" below.					ation. See		
I	ncompatibilities (Mat	terials to Avoid):						
	Microspheres wit photobleach. In st product may irreve	rong acid or base						
	Sodium Azide: So materials, oxidizing form explosive con	g materials. Sodi						
S	See Section IV: "Fire a	nd Explosion Haz	ard Data".					
I	Hazardous Decomposi	tion or Byprodu	cts:					
	Sodium Azide: Th	ermal decomposit	tion of sodium	azide may	form oxides of	nitrogen.		
I	Iazardous Polymeriza	ation	Wil	ll Occur		XV	Vill Not Occur	
SECT	TION VI. HEALTH	HAZARD DATA						
F	Route of Entry:	X Inl	nalation	X	Skin	X	Ingestion	
to	Health Hazards (Acut to the skin and mucou- luorescent dyes are sus	s membranes if in	ngested. The 1	toxic prope				
S	Sodium Azide: A loca odium azide by inhala genetic material.				•	1 -	•	1
	Eye Contact: Ey exposure may caus		um azide may	cause irri	tation with red	ness, pain, ar	nd blurred vision.	Chronic
	Skin Contact: Contact to the skin may cause irritation with redness and pain. Skin absorption may be fatal.							
	Inhalation: Vapors or fumes may cause irritation to the mucous membranes. Other symptoms may include a moderate reduction in blood pressure, variable pulse rate, headache, dizziness, fatigue, nausea, and faintness. Exposure to high concentrations may result in convulsions and death.							
	Ingestion: May be fatal if swallowed.							

MSDS 8640 Page 3 of 4

Target Organs: nerves, heart, and brain

Listed as a Carcinogen/Potential Carcinogen:

	1 03	110
In the National Toxicology Program (NTP) Report on Carcinogens		X
In the International Agency for Research on Cancer (IARC) Monographs	<u> </u>	X
By the Occupational Safety and Health Administration (OSHA)	<u> </u>	X

EMERGENCY AND FIRST AID PROCEDURES:

Sodium Azide:

Eye Contact: Flush eyes with plenty of water for at least 15 minutes. Obtain immediate medical attention.

Skin Contact: Wash skin with soap and copious amounts of water for at least 15 minutes. Remove contaminated clothing and shoes. Obtain medical assistance.

Ves

Inhalation: If adverse effects occur, remove to uncontaminated area. Give artificial respiration if not breathing by qualified personnel. If breathing is difficult, administer oxygen. Obtain immediate medical attention.

Ingestion: Obtain immediate medical attention. Drink plenty of water. **DO NOT** give anything by mouth to an unconscious or convulsive person.

SECTION VII. PRECAUTIONS FOR SAFE HANDLING AND USE

Microspheres with Immobilized Fluorescein Isothiocyanate:

Steps to be Taken in Case Material Is Released or Spilled: Wear vinyl gloves. Soak up spill with paper toweling, sand, or other non-combustible material, and rinse with water. Collect spilled material in an appropriate container for disposal. **DO NOT** dispose of material down the drain. Sodium azide forms explosive chemical compounds with lead and copper plumbing. If suspicious of or accidental drain disposal, flush with copious amounts of water to prevent azide build-up.

Waste Disposal: Follow all federal, state, and local regulations.

Handling and Storage: Good room ventilation is adequate for handling. If there is a possibility of inhalation exposure to dried particles, wear a NIOSH approved dust respirator.

NOTE: Contact lenses pose a special problem; soft lenses may absorb irritants and all lenses concentrate them. **DO NOT** wear contact lenses in the laboratory.

Store unopened bottles at temperatures between 2 °C and 6 °C in the dark in an upright position. Keep refrigerated. **DO NOT** freeze. Store and use in accordance with the Report of Investigation for SRM 8640.

SECTION VIII. SOURCE DATA/OTHER COMMENTS

Sources: Bangs Laboratories, Inc., MSDS Fluorescently Labeled Polymer Microspheres, 15 March 2002.

MDL Information Systems, Inc., MSDS *Phosphate Buffered Saline*, 15 December 2003.

MDL Information Systems, Inc., MSDS Polyoxyethylene (20) Sorbitan Monolaurate, 15 December 2003.

MDL Information Systems, Inc., MSDS Sodium Azide, 15 December 2003.

SRM 8640; *Microspheres with Immobilized Fluorescein Isothiocyanate*; National Institute of Standards and Technology, U.S. Department of Commerce: Gaithersburg, MD (2004).

Disclaimer: Physical and chemical data contained in this MSDS are provided only for use in assessing the hazardous nature of the material. The MSDS was carefully prepared, using current references; however, NIST does not certify the data in the MSDS. The certified value for this material is given in the NIST Certificate of Analysis.

MSDS 8640 Page 4 of 4